

2 α -Alkyl Gibberellins A₁ and A₄

Andrew M. Fowles, Michael H. Beale, David N. M. Jones, Jake MacMillan,* and Christine L. Willis

Department of Organic Chemistry, The University, Bristol BS8 1TS

An alkylcopper–magnesium complex, prepared from alkylmagnesium bromide, copper(I) iodide, and hexamethylphosphoric triamide in tetrahydrofuran, is described as a reagent for the *syn*-S_N2' displacement of the allylic lactonic oxygen in gibberellin A₃ methyl ester (13). The resultant 2 α -alkyl-1(10)-enes (24), (27), and (29) have been converted by known procedures into 2 α -methyl (7), 2 α -ethyl (9), and 2 α -propyl (11) derivatives of gibberellin A₁ (2). The 2 α -ethyl (5) and 2 β -ethyl (6) derivatives of gibberellin A₄ (1) have been prepared by 13-deoxygenation of the corresponding derivatives of gibberellin A₁ (2). The direct preparation of 2 α -alkyl derivatives of gibberellin A₄ by *syn*-S_N2' displacement of the allylic lactonic oxygen in the 3-tetrahydropyranyl ether of gibberellin A₇ methyl ester (15) with lithium dialkylcuprates works for methyl but not for higher alkyl groups. Attempts to prepare 2-methylgibberellin A₃ and 2-methylgibberellin A₇ gave the 2-methyl isomeric lactones (58) and (60).

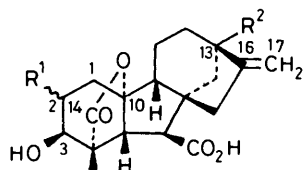
Continuing our studies on the effect of 2-alkyl substituents on the biological activity of gibberellins (GAs) we now describe the preparation of 2 α -alkyl derivatives of GA₄ (1) and GA₁ (2). As described in the preceding paper¹ the attempted preparation of 2 α -methyl- and 2 α -isopropyl-GA₄ by reduction of the 2 α -alkyl 3-ketones was unsuccessful, giving only the 2 α -alkyl 3 α -alcohols. Thus methods were required that introduced a 2 α -alkyl substituent with a 3 β -hydroxy group already in the molecule. Such methods have been developed, based upon the *syn*-S_N2' displacement of the allylic lactone in GA₃ methyl ester (13) using a novel alkylcopper–magnesium complex.

The reactions of GA₃ (14) and its methyl ester (13) with lithium dimethylcuprate have been shown² to give the *anti*-S_N2' products (22) and (23), respectively. In the case of GA₃ methyl ester (13), *ca.* 5% of the S_N2' product (31) was also isolated.³ However, Beale³ has shown that reaction of the 3-tetrahydropyranyl ether of GA₃ methyl ester (13) with lithium dimethylcuprate gave, after hydrolysis of the ether, a mixture of the 2 α - and 2 β -methyl derivatives (24) and (23). Applying this procedure to the 3-tetrahydropyranyl ether of GA₇ methyl ester (15), we obtained a 1:1 mixture of the 2 α - and 2 β -methyl derivatives (25) and (26). These compounds were then separately converted into 2 α - and 2 β -methyl-GA₄ (3) and (4) *via* the iodo lactones (32) and (33) and methyl esters (41) and (42), by the procedures described by MacMillan and Taylor.² The stereochemistry of 2 α -methyl-GA₄ (3) follows from the established⁴ stereochemistry of the known 2 β -methyl-GA₄ (4). However this stereochemistry was independently confirmed by using the previously established fact⁵ that the 16 α ,17-epoxide of GA₇ methyl ester (15) was catalytically [²H]hydrogenated to the 1 β ,2 β -[²H]₂epoxide of GA₄ methyl ester (43). Thus treatment of the 2 β -methyl iodo lactone (33) with 1,8-diazabicyclo[5.4.0]undec-7-ene gave 2-methyl GA₇ methyl ester (16). Conversion of this ester (16) into the corresponding 16 α ,17-epoxide followed by hydrogenation over palladium–calcium carbonate, then de-epoxidation and demethylation, afforded 2 α -methyl-GA₄ (3). The ¹H n.m.r. spectra of 2 α -methyl-GA₄ (3) and its precursors were unexceptional except for the 2 α -methyl-1(10)-ene (25). As for the corresponding product (24),³ obtained from the 3-tetrahydropyranyl ether of GA₃ methyl ester (13), the 6-H signal in the spectrum of the 2 α -methyl-1(10)-ene (25) occurred at an unusually high field (2.33 p.p.m.; *J* 9.3 Hz), and *J*_{2,3} was unusually large (9.0 Hz). These data were subsequently useful in distinguishing the 2 α - and 2 β -epimers of the corresponding 2 α - and 2 β -ethyl-1(10)-enes (27) and (28); they indicate a boat conformation for ring A in which

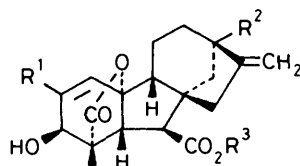
interaction between the 19-oic acid group and the 6-hydrogen atom is minimised.

Attempts to prepare 2 α -ethyl-GA₄ (5) by the method described for 2 α -methyl-GA₄ (3) were unsuccessful. Many products were formed in the reaction between the 3-tetrahydropyranyl ether of GA₇ methyl ester (15) and lithium diethylcuprate; the only isolated product, after hydrolysis of the tetrahydropyranyl ether, was the diene (57). The most likely explanation for the formation of the diene (57) is that lithium diethylcuprate undergoes β -elimination to a hydrido metal species which brings about S_N2' attack of hydride at C-2, followed by acid-catalysed dehydration during hydrolysis of the ether group.

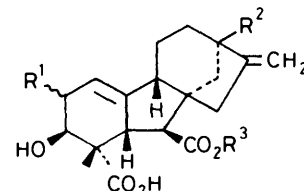
A general route to 2 α -alkyl-GAs was developed from a re-investigation of copper(I)-catalysed Grignard reactions with GA₃ methyl ester (13). Previously MacMillan and Taylor² showed that GA₃ methyl ester (13) reacted with methylmagnesium iodide in the presence of copper(I) bromide to give the *anti*-S_N2' product (23). However with ethylmagnesium halide and copper(I) halide no reaction occurred with GA₃ methyl ester (13)⁶ and, in the present work, several unidentified products were obtained in low yield from the 3-tetrahydropyranyl ether of GA₇ methyl ester (15). Nevertheless the Cu^I-catalysed S_N2' reaction of ethyl- (and other alkyl)-magnesium bromides with GA₃ methyl ester (13) occurred in the presence of hexamethylphosphoric triamide to give both *syn*- and *anti*-products in acceptable yields. After experimentation a general alkylcopper–magnesium reagent was developed from the alkylmagnesium bromide (2 mol equiv.), copper(I) iodide (1 mol equiv.) and hexamethylphosphoric triamide (1 mol equiv.) in tetrahydrofuran at –15 °C. This reagent was then treated with GA₃ methyl ester (13) (0.5–1.0 mol equiv.) in tetrahydrofuran at –40 °C. Under these conditions, ethylmagnesium bromide gave the 2 α -ethyl- and 2 β -ethyl-1(10)-enes (27) and (28) in the ratio *ca.* 1:1, in 47% overall yield; propylmagnesium bromide gave the 2 α -propyl- and 2 β -propyl-1(10)-enes (29) and (30) in the ratio 3:1 in an overall yield of 41%; and methylmagnesium bromide gave 2 α -methyl- and 2 β -methyl-1(10)-enes (24) and (23) in the ratio 3:1 in an overall yield of 70%. It is important that GA₃ methyl ester (13) is added to the alkylcopper reagent at –40 °C and that the reaction temperature is kept below –15 °C. The nature of the alkylcopper–magnesium reagent is not known, but it is presumably solvated and sterically demanding, to account for the high stereoselectivity for the 2 α -alkylation products. The reagent is violet like the solid (Ph₂CuPh₂Mg·*n*THF), prepared



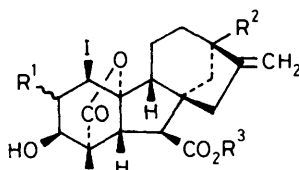
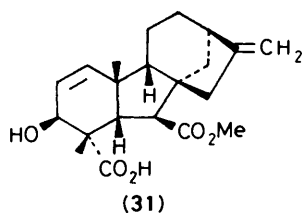
- (1) $R^1 = R^2 = H$
 (2) $R^1 = H, R^2 = OH$
 (3) $R^1 = \alpha\text{-Me}, R^2 = H$
 (4) $R^1 = \beta\text{-Me}, R^2 = H$
 (5) $R^1 = \alpha\text{-Et}, R^2 = H$
 (6) $R^1 = \beta\text{-Et}, R^2 = H$
 (7) $R^1 = \alpha\text{-Me}, R^2 = OH$
 (8) $R^1 = \beta\text{-Me}, R^2 = OH$
 (9) $R^1 = \alpha\text{-Et}, R^2 = OH$
 (10) $R^1 = \beta\text{-Et}, R^2 = OH$
 (11) $R^1 = \alpha\text{-Pr}, R^2 = OH$
 (12) $R^1 = \beta\text{-Pr}, R^2 = OH$



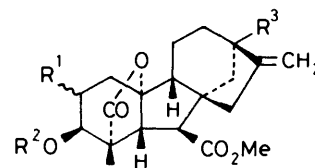
- (13) $R^1 = H, R^2 = OH, R^3 = Me$
 (14) $R^1 = R^3 = H, R^2 = OH$
 (15) $R^1 = R^2 = H, R^3 = Me$
 (16) $R^1 = R^3 = Me, R^2 = H$
 (17) $R^1 = R^2 = R^3 = H$
 (18) $R^1 = Et, R^2 = OH, R^3 = Me$
 (19) $R^1 = Me, R^2 = OH, R^3 = H$
 (20) $R^1 = Et, R^2 = OH, R^3 = H$
 (21) $R^1 = Me, R^2 = R^3 = H$



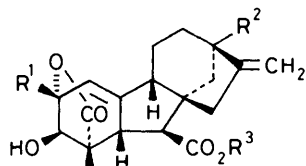
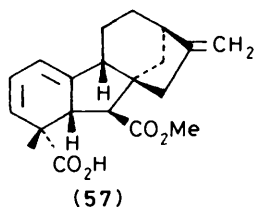
- (22) $R^1 = \beta\text{-Me}, R^2 = OH, R^3 = H$
 (23) $R^1 = \beta\text{-Me}, R^2 = OH, R^3 = Me$
 (24) $R^1 = \alpha\text{-Me}, R^2 = OH, R^3 = Me$
 (25) $R^1 = \alpha\text{-Me}, R^2 = H, R^3 = Me$
 (26) $R^1 = \beta\text{-Me}, R^2 = H, R^3 = Me$
 (27) $R^1 = \alpha\text{-Et}, R^2 = OH, R^3 = Me$
 (28) $R^1 = \beta\text{-Et}, R^2 = OH, R^3 = Me$
 (29) $R^1 = \alpha\text{-Pr}, R^2 = OH, R^3 = Me$
 (30) $R^1 = \beta\text{-Pr}, R^2 = OH, R^3 = Me$



- (32) $R^1 = \alpha\text{-Me}, R^2 = H, R^3 = Me$
 (33) $R^1 = \beta\text{-Me}, R^2 = H, R^3 = Me$
 (34) $R^1 = \alpha\text{-Me}, R^2 = OH, R^3 = Me$
 (35) $R^1 = \alpha\text{-Et}, R^2 = OH, R^3 = Me$
 (36) $R^1 = \beta\text{-Et}, R^2 = OH, R^3 = Me$
 (37) $R^1 = \alpha\text{-Pr}, R^2 = OH, R^3 = Me$
 (38) $R^1 = \beta\text{-Pr}, R^2 = OH, R^3 = Me$
 (39) $R^1 = \beta\text{-Me}, R^2 = OH, R^3 = H$
 (40) $R^1 = \beta\text{-Me}, R^2 = OH, R^3 = SnBu_3$



- (41) $R^1 = \alpha\text{-Me}, R^2 = R^3 = H$
 (42) $R^1 = \beta\text{-Me}, R^2 = R^3 = H$
 (43) $R^1 = R^2 = R^3 = H$
 (44) $R^1 = \alpha\text{-Me}, R^2 = H, R^3 = OH$
 (45) $R^1 = \alpha\text{-Et}, R^2 = H, R^3 = OH$
 (46) $R^1 = \beta\text{-Et}, R^2 = H, R^3 = OH$
 (47) $R^1 = \alpha\text{-Pr}, R^2 = H, R^3 = OH$
 (48) $R^1 = \beta\text{-Pr}, R^2 = H, R^3 = OH$
 (49) $R^1 = \alpha\text{-Et}, R^2 = Ac, R^3 = OH$
 (50) $R^1 = \beta\text{-Et}, R^2 = Ac, R^3 = OH$
 (51) $R^1 = \alpha\text{-Et}, R^2 = Ac, R^3 = OCOC_2Me$
 (52) $R^1 = \beta\text{-Et}, R^2 = Ac, R^3 = OCOC_2Me$
 (53) $R^1 = \alpha\text{-Et}, R^2 = Ac, R^3 = H$
 (54) $R^1 = \beta\text{-Et}, R^2 = Ac, R^3 = H$
 (55) $R^1 = \alpha\text{-Et}, R^2 = R^3 = H$
 (56) $R^1 = \beta\text{-Et}, R^2 = R^3 = H$



- (58) $R^1 = Me, R^2 = OH, R^3 = H$
 (59) $R^1 = Et, R^2 = OH, R^3 = Me$
 (60) $R^1 = Me, R^2 = R^3 = H$
 (61) $R^1 = R^3 = Me, R^2 = H$

by Costa *et al.*⁷ from phenylmagnesium bromide and copper(I) bromide in tetrahydrofuran.

By previously described methods,² the 2 α - and 2 β -alkyl-1(10)-enes (24) and (27)—(30) were converted into the 2 α - and 2 β -alkyl esters (44)—(48) *via* the iodo lactones (34)—(38). Demethylation of the methyl esters then gave 2 α -methyl-GA₁ (7), 2 α -ethyl-GA₁ (9), 2 α -propyl-GA₁ (11), and 2 β -propyl-GA₁ (12). This sequence of reactions was not applied to the

2 β -methyl-1(10)-ene (23), since the preparation of 2 β -methyl-GA₁ (8) by this route has been described² previously. Similarly since 2 β -ethyl-GA₁ (10) has also been previously prepared,² the 2 β -ethyl-GA₁ methyl ester (46) was retained for conversion into 2 β -ethyl-GA₄ (6) as described later. The stereochemistry of the 2 α - and 2 β -alkyl groups in all the compounds was assigned from their ¹H n.m.r. spectra and from the results of treatment of the iodo lactones (34)—(36) with 1,8-diazabicyclo[5.4.0]undec-7-

ene in refluxing toluene. As expected the 2 α -alkyl-1 β -iodo lactones (34) and (35) did not react, but the 2 β -ethyl-1 β -iodo lactone (36) underwent *trans*-dehydroiodination to give a mixture of 2-ethyl-GA₃ methyl ester (18) and the corresponding 19,2-isomeric lactone (59). The formation of the isomeric lactone (59) is discussed later.

2 α -Ethyl-GA₄ (5) and 2 β -ethyl-GA₄ (6) were prepared from 2 α - and 2 β -ethyl-GA₁ methyl esters (45) and (46). 13-Deoxygenation⁸ of the 3-acetyl 13-oxalyl esters (51) and (52) with tributylstannane gave the 3-acetates (53) and (54), which were hydrolysed in two steps *via* the methyl esters (55) and (56) to 2 α - and 2 β -ethyl-GA₄ (5) and (6).

Attempts to prepare 2-methyl-GA₃ (19), 2-ethyl-GA₃ (20), and 2-methyl-GA₇ (21) were unsuccessful. 1 β -Iodo-2 β -methyl-GA₁ (39), prepared from GA₃ (14) by the method of MacMillan and Taylor,² was converted into the tributylstannyl ester (40) and treated with 1,8-diazabicyclo[5.4.0]undec-7-ene in refluxing toluene. After hydrolysis of the stannyl ester, the isomeric 19,2-lactone (58) of 2-methyl-GA₃ (19) was obtained as the sole product. As noted earlier, treatment of 1 β -iodo-2 β -ethyl-GA₁ methyl ester (36) with 1,8-diazabicyclo[5.4.0]undec-7-ene gave a mixture of 2-ethyl-GA₃ methyl ester (18) and the isomeric lactone (59). Similarly, demethylation of 2-methyl-GA₇ methyl ester (16), obtained as described earlier, with sodium propane-1-thiolate in hexamethylphosphoric triamide, yielded the isomeric lactone (60) and no 2-methyl-GA₇ (21). Since 19,10 \rightarrow 19,2-lactone isomerisation in aqueous alkali occurs *via* a 2 β ,3 β -epoxide,⁹ the 3 β -hydroxy group of 2-methyl-GA₃ methyl ester was protected by formation of the 3-tetrahydropyranyl ether. However hydrolysis of the methyl ester of this tetrahydropyranyl ether with sodium hydroxide in aqueous methanol, followed by acid hydrolysis of the ether, again gave the isomeric lactone (58) as the sole product. This latter result suggested that isomerisation of the lactone can also occur under acidic conditions, and this was demonstrated by conversion of 2-methyl-GA₇ methyl ester (16) into the isomeric lactone (61) by treatment with toluene-*p*-sulphonic acid in either dichloromethane or acetone-methanol (1:1). The mechanism of this acid-catalysed lactone isomerisation is not known but the ready rearrangement of the lactone system in 2-alkyl derivatives of GA₃ and GA₇ is probably facilitated by the electron-donating 2-alkyl groups.

The biological activities of the 2 β - and 2 α -alkyl derivatives described in this and the preceding paper will be presented elsewhere. However these results show that 2 α -methyl and 2 α -ethyl substituents confer high biological activity of GA₄ (1) and GA₁ (2).

Experimental

General experimental details and the preparation of tributylstannane, sodium propanethiolate, and lithium dialkylcuprate solutions are given in the preceding paper.¹

Lithium Dialkylcuprate Reactions.—(a) *Lithium dimethylcuprate and 3-O-tetrahydropyranylgibberellin A₇ methyl ester.* A mixture (657 mg) containing 24% GA₄ (1) and 73% GA₇ (17) in methanol was treated with ethereal diazomethane. The crystalline mixture of the methyl esters (15) and (43) obtained on evaporation was stirred with 2,3-dihydropyran (0.35 ml; freshly distilled from potassium hydroxide) and a few crystals of toluene-*p*-sulphonic acid in dry dichloromethane (10 ml). After 3.5 h the solution was evaporated under reduced pressure and the resultant gum in dry dichloromethane (4 ml) was added to a solution of lithium dimethylcuprate (3.8 mmol) in dry diethyl ether (20 ml). The mixture was stirred at 0 °C for 1 h, then at room temperature for 3 h. Work-up gave a gum which was stirred in acetone (27 ml) and methanol (3 ml) containing a few

crystals of toluene-*p*-sulphonic acid. After 4 h the mixture was evaporated under reduced pressure and the residue in ethyl acetate (30 ml) was extracted with saturated aqueous sodium hydrogen carbonate (3 \times 30 ml). The aqueous fractions were acidified to pH 2 and extracted with ethyl acetate (3 \times 30 ml). Evaporation under reduced pressure followed by flash chromatography with ethyl acetate-light petroleum-acetic acid (50:150:1 then 60:140:1) gave, sequentially, 2 β -methyl-1,10-didehydro-GA₄ 7,19-diacid 7-methyl ester (26) as a gum (81 mg) (Found: M^+ , 360.1925. C₂₁H₂₈O₅ requires M , 360.1936); δ 1.11 (3 H, d, J 7.3 Hz, 2 β -Me), 1.34 (3 H, s, 18-H₃), 3.04 (1 H, br m, 5-H), 3.13 (1 H, d, J 6.3 Hz, 6-H), 3.70 (3 H, s, OMe), 3.87 (1 H, d, J 2.7 Hz, 3 α -H), 4.89 (2 H, br s, 17-H₂ or 17-H and 1-H), and 4.99 (1 H, br s, 17- or 1-H); m/z 360 (M^+ , 4%), 342 (26), 328 (4), 300 (15), 283 (12), 282 (17), 238 (21), 237 (100), 195 (11), 169 (12), 119 (10), and 9 (13); and 2 α -methyl-1,10-didehydro-GA₄ 7,19-diacid 7-methyl ester (25) (88 mg) (Found: M^+ , 360.1956. C₂₁H₂₈O₅ requires M , 360.1936); δ 1.18 (3 H, d, J 6.8 Hz, 2 α -Me), 1.34 (3 H, s, 18-H₃), 2.38 (1 H, d, J 9.3 Hz, 6-H), 3.00 (1 H, br d, J 9.4 Hz, 5-H), 3.52 (1 H, d, J 9.0 Hz, 3 α -H), 3.67 (3 H, s, OMe), 4.89 (2 H, br s, 17-H₂ or 17-H and 1-H), and 5.07 (1 H, br s, 17- or 1-H); m/z 360 (M^+ , 4%), 342 (11), 328 (7), 300 (10), 296 (22), 283 (17), 282 (28), 237 (82), 169 (16), and 91 (20).

(b) *Lithium diethylcuprate and 3-O-tetrahydropyranylgibberellin A₇ methyl ester.* To lithium (1.0 g, sliced strips) in dry diethyl ether (30 ml) at 0 °C under nitrogen was added 1/8th of a solution of ethyl bromide (5.3 ml; distilled from phosphorus pentaoxide) in dry diethyl ether (20 ml). When attack of the lithium was established the remainder of the ethereal ethyl bromide was added dropwise over 2 h. The cloudy solution was allowed to settle overnight at -20 °C. Titration against diphenylacetic acid¹⁰ indicated the ethyl-lithium concentration to be 0.33M. A portion (27 ml) of this solution was added to a stirred suspension of copper(I) iodide (840 mg) in dry tetrahydrofuran (20 ml) at -20 °C under nitrogen. To the resultant maroon-black solution was added a mixture of the tetrahydropyranyl ethers of GA₄ and GA₇ methyl esters, (43) and (15), containing 0.75 mmol GA₇ derivative and prepared as described in (a), in tetrahydrofuran (5 ml). The temperature was kept between -20 and 0 °C for 1 h and the mixture was then allowed to warm to room temperature. After 4 h (total reaction time) work-up and cleavage of the tetrahydropyranyl ether as described in (a) gave GA₄ methyl ester (43) (125 mg), identical (¹H n.m.r. and t.l.c.) with an authentic sample, and a gum, believed to be 20-norgibberella-1(10),2,16-triene-7,19-dioic acid 7-methyl ester (57) (*ca.* 40 mg) as a gum: δ 1.34 (3 H, s, 18-H₃), 2.98 (1 H, d, J 4.6 Hz, 6-H), 3.36 (1 H, br m, 5-H), 3.71 (3 H, s, OMe), 4.89 (2 H, br s, 17-H₂), 5.37 (1 H, d, J 9.5 Hz, 3-H), 5.63 (1 H, ddd, J 5.2, 2.6, and 2.6 Hz, 1-H), and 6.12 (1 H, dd, J 9.6 and 5.2 Hz, 2-H); m/z 328 (M^+ , 7%), 313 (16), 282 (19), 281 (19), 223 (100), 222, (59), and 155 (27).

Reactions of Gibberellin A₃ Methyl Ester (13) and Alkylcopper-Magnesium Reagents.—(a) *Methylcopper-magnesium reagent.* To copper(I) iodide (338 mg) slurried in tetrahydrofuran (20 ml) under nitrogen was added hexamethylphosphoric triamide (0.31 ml), and the mixture was cooled to -15 °C. Dropwise addition of methylmagnesium bromide (1.16 ml; 3M solution in diethyl ether) gave a purple supernatant and a yellow solid. The mixture was cooled to -40 °C and GA₃ methyl ester (13) (314 mg) was added in tetrahydrofuran (6 ml). The temperature was allowed to rise and the mixture was stirred for 1 h at between -25 and -15 °C and then for 1 h at room temperature. After work-up the ethyl acetate extracts were washed with saturated aqueous sodium hydrogen carbonate (\times 4). Evaporation of the ethyl acetate fraction under reduced pressure afforded crude starting material (13) (84 mg). Acidification of the aqueous washings to pH 2 and extraction

with ethyl acetate ($\times 5$) gave a gum. Flash chromatography with ethyl acetate–light petroleum–acetic acid (50:50:1) yielded, sequentially, the known *ent*-3 α ,13-dihydroxy-2 α -methyl-20-norgibberell-1(10),16-diene-7,19-dioic acid 7-methyl ester (**23**) (54 mg), identified by ^1H n.m.r. and mass spectrometry; and *ent*-3 α ,13-dihydroxy-2 β -methyl-20-norgibberella-1(10),16-diene-7,19-dioic acid 7-methyl ester (**24**) as a gum (177 mg) (Found: M^+ , 376.1869; $M^+ - 18$, 358.1768. $\text{C}_{21}\text{H}_{28}\text{O}_6$ requires M , 376.1886; $M - 18$, 358.1780); δ 1.17 (3 H, d, J 6.8 Hz, 2 α -Me), 1.36 (3 H, s, 18-H₃), 2.39 (d, J 8.3 Hz, 6-H; superimposed on other signals), 3.01 (1 H, br d, J 8.3 Hz, 5-H), 3.52 (1 H, d, J 8.3 Hz, 3 α -H), 3.68 (3 H, s, OMe), 4.97 (1 H, br s, 17- or 1-H), and 5.11 (2 H, br s, 17-H₂ or 17-H and 1-H); m/z 376 (M^+ , 6%), 358 (24), 317 (30), 316 (92), 312 (39), 298 (54), 254 (32), 253 (100), and 91 (34).

(b) *Ethylcopper–magnesium reagent*. To copper(I) iodide (2.56 g), slurried in dry tetrahydrofuran under nitrogen, was added hexamethylphosphoric triamide (2.4 ml), and the mixture was cooled to -15°C . Dropwise addition of ethylmagnesium bromide (8.9 ml; 3M solution in diethyl ether) gave a purple solution (on a smaller scale slight warming may be required to form the complex). After cooling to -40°C and stirring at this temperature for 5 min, GA₃ methyl ester (**13**) (1.73 g) in dry tetrahydrofuran (9 ml) was added slowly. The reaction mixture thickened markedly but was sheared by rapid stirring. The temperature was allowed to rise to -15°C and stirring was continued for 3 h (the mixture became dark purple). Work-up as in (a) gave starting material (**13**) (284 mg) and an acidic gum which was subjected to flash chromatography with ethyl acetate–light petroleum (55:45) to give sequentially: (i) *ent*-2 α -ethyl-3 α ,13-dihydroxy-20-norgibberella-1(10),16-diene-7,19-dioic acid 7-methyl ester (**28**) (496 mg) as a gum (Found: $M^+ - 18$, 372.1943. $\text{C}_{22}\text{H}_{30}\text{O}_6$ requires $M - 18$, 372.1937); δ [(CD_3)₂CO] 1.00 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.29 (3 H, s, 18-H₃), 3.07 (1 H, br m, 5-H), 3.21 (1 H, d, J 6.1 Hz, 6-H), 3.67 (3 H, s, OMe), 4.01 (1 H, d, J 2.2 Hz, 3 α -H), 4.90 (1 H, br s, 17-H), and 5.07 (2 H, br s, 17- and 1-H) [^1H n.m.r. homonuclear decoupling: on irradiation at 3.07 p.p.m. the signal at 3.15 p.p.m. collapsed to a singlet; on irradiation at 3.21 p.p.m. the signal at 3.07 p.p.m. became a broad singlet; on irradiation of the singlet at 5.07 p.p.m. the signal at 3.07 p.p.m. sharpened (*i.e.* some coupling lost)]; m/z 390 (M^+ , 2%), 372 (32), 340 (17), 331 (10), 330 (10), 313 (11), 312 (15), 297 (12), 268 (27), 267 (100), 239 (12), and 91 (14); δ (7,19-dimethyl ester) 1.02 (3 H, t, J 7.3 Hz, CH₂Me), 1.31 (3 H, s, 18-H₃), 3.01 (1 H, br m, 5-H), 3.15 (1 H, d, J 6.1 Hz, 6-H), 3.64 (3 H, s, OMe), 3.70 (3 H, s, OMe), 3.96 (1 H, br s, 3 α -H), 4.98 (1 H, br s, 17-H), 5.06 (1 H, br s, 1-H), and 5.11 (1 H, br s, 17-H); (ii) *ent*-2 β -ethyl-3 α ,13-dihydroxy-20-norgibberella-1(10),16-diene-7,19-dioic acid 7-methyl ester (**27**) (427 mg) as a gum (Found: $M^+ - 18$, 372.1925. $\text{C}_{22}\text{H}_{30}\text{O}_6$ requires $M - 18$, 372.1937); δ [(CD_3)₂CO] (7-methyl ester 19-acid) 0.99 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.32 (3 H, s, 18-H₃), 2.56 (1 H, d, J 7.8 Hz, 6-H), 3.00 (1 H br d, J 8.1 Hz, 5-H), 3.63 (3 H, s, OMe), 4.88 (1 H, br s, 17-H), 5.08 (1 H, br s, 17-H), and 5.21 (1 H, ddd, J 2.2 Hz, $\times 3$, 1-H) [^1H n.m.r. homonuclear decoupling: on irradiation at 2.56 p.p.m. the signal at 3.00 p.p.m. collapsed to a broad singlet; on irradiation at 3.00 p.p.m. the signal at 2.56 p.p.m. collapsed to a sharp singlet and the signal at 5.25 p.p.m. collapsed to a broad singlet (triplet substructure, J ca. 2 Hz); on irradiation at 5.25 p.p.m. the signal at 3.00 p.p.m. became a ddd ($J_{5,6}$ 7.6, $J_{5,9}$ 2.1, $J_{5,2}$ 0.5 Hz)]; m/z 390 (M^+ , 4%), 372 (21), 331 (22), 330 (68), 326 (25), 313 (18), 312 (37), 268 (25), 267 (100), 239 (30), and 91 (30); δ (7,19-dimethyl ester) 0.99 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.35 (3 H, s, 18-H₃), 2.18 (1 H, d, J 8.1 Hz, 6-H), 3.05 (1 H, br dt, $J_{5,6}$ 7.8 Hz, $J_{5,1} \approx J_{5,9} = 2.0$ Hz, 5-H), 3.62 (3 H, s, OMe), 3.65 (1 H, d, J 6.4 Hz, 3 α -H), 3.71 (3 H, s, OMe), 4.97 (1 H, br s, 17-H), 5.12 (1 H, br s, 17-H), and 5.20 (1 H, ddd, $J \approx 2.0$ Hz $\times 3$, 1-H).

On a smaller scale (*ca.* 0.33 of above) repetition afforded 2 β -

ethyl- and 2 α -ethyl-1,10-didehydro-GA₁ 7,19-diacid 7-methyl esters (**28**) and (**27**) in the ratio 36:64.

(c) *Propylcopper–magnesium reagent*. The procedures described in (b) were followed with the following quantities: copper(I) iodide (1.12 g) in tetrahydrofuran (40 ml) containing hexamethylphosphoric triamide (1 ml); 2M propylmagnesium chloride in diethyl ether (6 ml); and gibberellin A₃ methyl ester (**13**) (1.06 g) in tetrahydrofuran (10 ml). Starting material (**13**) (258 mg) was recovered in the neutral fraction. The acidic product was subjected to flash chromatography using ethyl acetate–light petroleum–acetic acid (55:45:1) to give, sequentially: (i) *ent*-3 α ,13-dihydroxy-2 α -propyl-20-norgibberella-1(10),16-diene-7,19-dioic acid 7-methyl ester (**30**) as a gum (193 mg) (Found: M^+ , 404.2183. $\text{C}_{23}\text{H}_{32}\text{O}_6$ requires M , 404.2199); δ 0.97 (3 H, t, J 6.1 Hz, 2-CH₂CH₂Me), 1.33 (3 H, s, 18-H₃), 3.12 (1 H, d, J 6.1 Hz, 6-H), 3.71 (3 H, s, OMe), 3.95 (1 H, br s, 3 α -H), 4.94 (1 H, br s, 17-H), and 5.06 (2 H, br s, 17- and 1-H); m/z 404 (M^+ , 2%), 386 (35), 354 (15), 345 (9), 344 (9), 327 (10), 326 (14), 311 (10), 282 (23), 281 (100), 239 (8), 105 (7), and 91 (9); and (ii) *ent*-3 α ,13-dihydroxy-2 β -propyl-20-norgibberella-1(10),16-diene-7,19-dioic acid 7-methyl ester (**29**) (300 mg) as a gum (Found: M^+ , 404.2208. $\text{C}_{23}\text{H}_{32}\text{O}_6$ requires M , 404.2199); δ 0.96 (3 H, t, J 6.4 Hz, 2-CH₂CH₂Me), 1.36 (3 H, s, 18-H₃), 3.61 (1 H, superimposed on signal at 3.67 p.p.m., 5-H), 3.67 (3 H, s, OMe₃), 3.72 (1 H, d, J 7.1 Hz, 3 α -H), 4.97 (1 H, br s, 17-H), 5.12 (1 H, br s, 17-H), and 5.22 (1 H, br s, 1-H); m/z 404 (M^+ , 5%), 386 (20), 345 (27), 344 (63), 340 (30), 327 (19), 326 (41), 282 (26), 281 (100), 239 (24), 105 (15), and 91 (26).

Iodo Lactones.—(a) *ent*-3 α ,10 β -Dihydroxy-1 α -iodo-2 α -methyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (**33**). The 2 β -methyl-1(10),16-diene-7,19-dioic acid 7-methyl ester (**23**) (937 mg) in tetrahydrofuran (25 ml) and dichloromethane (40 ml) was stirred vigorously with saturated aqueous sodium hydrogen carbonate (40 ml) and iodine (1.67 g) for 1 h. The organic phase was washed with saturated aqueous sodium thiosulphate (2×50 ml) followed by water (2×30 ml). After drying (MgSO₄), evaporation under reduced pressure gave 1 β -iodo-2 β -methyl-GA₄ methyl ester (**33**) (1.06 g), m.p. 137–140 $^\circ\text{C}$ (from ethyl acetate–light petroleum) (Found: M^+ , 486.0875. $\text{C}_{21}\text{H}_{27}\text{IO}_5$ requires M , 486.0905); δ 1.18 (3 H, s, 18-H₃), 1.26 (3 H, d, J 6.6 Hz, 2 β -Me), 2.64 (1 H, d, J 10.3 Hz, 6-H), 3.70 (2 H, m, 3 β -OH and 3 α -H), 3.73 (3 H, s, OMe), 3.77 (1 H, d, J 10.5 Hz, 5-H), 4.39 (1 H, d, J 4.9 Hz, 1 α -H), 4.90 (1 H, br s, 17-H), and 5.00 (1 H, br s, 17-H); δ [(CD_3)₂CO] 1.10 (3 H, s, 18-H₃), 1.20 (3 H, d, J 6.6 Hz, 2 β -Me), 2.63 (d, J 10.4 Hz, 6-H; superimposed on other signals), 3.63 (1 H, br m, 3 α -H), 3.71 (3 H, s, OMe), 3.87 (1 H, d, J 10.5 Hz, 5-H), 4.61 (2 H, d, J 5.1 Hz, 1-H and 3 β -OH), 4.93 (1 H, br s, 17-H), and 5.03 (1 H, br s, 17-H); m/z 486 (M^+ , 17%), 454 (62), 426 (37), 359 (12), 341 (11), 327 (33), 313 (25), 299 (43), 237 (100), 105 (18), and 91 (30).

(b) *ent*-3 α ,10 β -Dihydroxy-1 α -iodo-2 β -methyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (**32**). The 2 α -methyl-1(10),16-diene-7,19-dioic acid 7-methyl ester (**25**) (71 mg) in tetrahydrofuran (8 ml) and dichloromethane (10 ml) was stirred vigorously with saturated aqueous sodium hydrogen carbonate (10 ml) and iodine (125 mg). After 1 h the mixture was worked up as described in (a). Flash chromatography with ethyl acetate–light petroleum (30:70) gave 1 β -iodo-2 α -methyl-GA₄ methyl ester (**32**), needles (54 mg; low yield through spillage during work-up), m.p. 142–143 $^\circ\text{C}$ (from ethyl acetate–light petroleum) (Found: C, 51.7; H, 5.9. $\text{C}_{21}\text{H}_{27}\text{IO}_5$ requires C, 51.9; H, 5.6%); δ 1.16 (3 H, d, J 8.3 Hz, 2 α -Me), 1.21 (3 H, s, 18-H₃), 2.34 (1 H, d, J 5.6 Hz, 3 β -OH), 2.69 (1 H, d, J 10.3 Hz, 6-H), 3.66 (1 H, d, J 5.6 Hz, 3 α -H), 3.74 (3 H, s, OMe), 3.80 (1 H, d, J 10.3 Hz, 5-H), 4.21 (1 H, s, 1 α -H), 4.94 (1 H, br s, 17-H), and 5.00 (1 H, br s, 17-H); m/z 486 (M^+ , 6%), 454 (12), 359 (12), 341 (11), 327 (43), 313 (17), 299 (74), 281 (43), 237 (100), 105 (17), and 91 (28).

(c) ent-3 α ,10 β ,13-Trihydroxy-1 α -iodo-2 β -methyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (34). The 2 α -methyl-1(10),16-diene 7,19-diacid 7-methyl ester (24) (175 mg) in tetrahydrofuran (5 ml) and dichloromethane (10 ml) was vigorously stirred with saturated aqueous sodium hydrogen carbonate (10 ml) and iodine (146 mg) for 1 h. Work-up as described in (a) gave 1 β -iodo-2 α -methyl-GA₁ methyl ester (34), prisms (169 mg), m.p. 185 °C (decomp.) (from acetone-tetrahydrofuran-light petroleum) (Found: C, 49.9; H, 5.3%; M⁺, 502.0833. C₂₁H₂₇O₆ requires C, 50.2; H, 5.4%; M, 502.0854); δ (C₅D₅N) 1.25 (3 H, d, J 8.1 Hz, 2 α -Me), 1.55 (3 H, s, 18-H₃), 3.04 (1 H, t, J 8.5 Hz, 9-H; under signal at 3.08 p.p.m.), 3.08 (1 H, d, J 9.5 Hz, 6-H; superimposed on signal at 3.04 p.p.m.), 3.26 (1 H, q, J 8.2 Hz, 2 β -H), 3.63 (3 H, s, OMe), 4.03 (1 H, s, 3 α -H), 4.47 (1 H, d, J 9.3 Hz, 5-H), 4.67 (1 H, s, 1 α -H), 5.10 (1 H, br s, 17-H), and 5.61 (1 H, br s, 17-H); m/z 502 (M⁺, 64%), 470 (38), 443 (98), 375 (28), 357 (79), 343 (87), 329 (100), 315 (72), 297 (66), 269 (68), 253 (93), 135 (76), 91 (69), 69 (69), and 43 (93).

(d) ent-2 α -Ethyl-3 α ,10 β ,13-trihydroxy-1 α -iodo-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (36). The 2 β -ethyl-1(10),16-diene 7,19-diacid 7-methyl ester (28) (496 mg) in tetrahydrofuran (10 ml) and dichloromethane (20 ml) was stirred vigorously with saturated aqueous sodium hydrogen carbonate (20 ml) and iodine (355 mg) for 1 h. Work-up as described in (a) gave 2 β -ethyl-1 β -iodo-GA₁ methyl ester (36) (637 mg) as a gum (Found: M⁺, 516.0997. C₂₂H₂₉O₆ requires M, 516.1011); δ 0.94 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.19 (3 H, s, 18-H₃), 2.68 (1 H, d, J 9.8 Hz, 6-H; superimposed on the signal at 2.78 p.p.m.), 2.78 (1 H, br t, 9-H; under signal at 2.68 p.p.m.), 3.74 (4 H, s, OMe and 3 α -H), 3.77 (1 H, d, J 9.8 Hz, 5-H), 4.36 (1 H, d, J 3.9 Hz, 1-H), 5.01 (1 H, br s, 17-H), and 5.28 (1 H, br s, 17-H); δ [(CD₃)₂CO] 0.93 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.11 (3 H, s, 18-H₃), 2.62 (1 H, d, J 9.8 Hz, 6-H), 2.79 (1 H, br t, 9-H), 3.72 (3 H, s, OMe), 3.88 (1 H, d, J 10.0 Hz, 5-H), 3.90 (1 H, br s, 3 α -H), 4.58 (2 H, d, J 5.1 Hz, 1-H and 3 β -OH), 4.93 (1 H, br s, 17-H), and 5.22 (1 H, br s, 17-H); m/z 516 (M⁺, 39%), 457 (66), 389 (14), 371 (61), 357 (28), 344 (17), 343 (60), 339 (18), 329 (33), 325 (18), 311 (45), 283 (47), 267 (81), 265 (34), and 71 (100).

(e) ent-2 β -Ethyl-3 α ,10 β ,13-trihydroxy-1 α -iodo-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (35). The 2 α -ethyl-1(10),16-diene 7,19 diacid 7-methyl ester (27) (332 mg) in tetrahydrofuran (15 ml) and dichloromethane (20 ml) was stirred vigorously with saturated aqueous sodium hydrogen carbonate (20 ml) and iodine (237 mg) for 1 h. Work-up as described in (a) gave 2 α -ethyl-1 β -iodo-GA₁ methyl ester (35) (440 mg), m.p. 140–145 °C (decomp.) (from acetone-ethyl acetate-light petroleum) (Found: C, 51.1; H, 5.6; I, 24.4. C₂₂H₂₉O₆ requires C, 51.2; H, 5.6; I, 24.6%); δ 0.99 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.21 (3 H, s, 18-H₃), 2.66 (2 H, m; under signal at 2.68 p.p.m.), 2.68 (1 H, d, J 9.5 Hz, 6-H; superimposed on signal at 2.66 p.p.m.), 3.75 (4 H, s, OMe and 3 α -H), 3.83 (1 H, d, J 9.5 Hz, 5-H), 4.30 (1 H, s, 1 α -H), 5.00 (1 H, br s, 17-H), and 5.27 (1 H, br s, 17-H); δ [(CD₃)₂CO] 0.97 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.15 (3 H, s, 18-H₃), 2.62 (1 H, d, J 9.3 Hz, 6-H; superimposed on signal at 2.66 p.p.m.), 2.66 (1 H, t, 9-H; under signal at 2.62 p.p.m.), 3.73 (4 H, s, OMe and 3 α -H), 3.90 (1 H, s, 13-OH; superimposed on upfield arm of signal at 3.91 p.p.m.), 3.91 (1 H, d, J ca. 9 Hz, 5-H; upfield arm is superimposed on signal at 3.90 p.p.m.), 4.50 (1 H, s, 1 α -H), 4.80 (1 H, d, J 4.2 Hz, 3 β -OH), 4.93 (1 H, br s, 17-H), and 5.20 (1 H, br s, 17-H) [on addition of D₂O the signals at 4.80 p.p.m. (3 β -OH) and 3.90 p.p.m. (13-OH) disappeared, that of 3 α -H collapsed to 1 H, s at 3.72 p.p.m. and that of 5-H became obvious as 1 H, d, J 9.3 Hz, at 3.87 p.p.m.]; m/z 516 (M⁺, 47), 484 (25), 457 (95), 389 (33), 371 (93), 357 (85), 343 (100), 339 (39), 330 (67), 325 (36), 311 (79), 283 (61), and 267 (73).

(f) ent-3 α ,10 β ,13-Trihydroxy-1 α -iodo-2 α -propyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone

(38). The 2 β -propyl-1(10),16-diene-7,19-dioic acid 7-methyl ester (30) (232 mg) in tetrahydrofuran (20 ml) and dichloromethane (40 ml) was stirred vigorously with saturated aqueous sodium hydrogen carbonate (40 ml) and iodine (161 mg) for 1 h. Work-up as in (a) gave 1 β -iodo-2 β -propyl-GA₁ methyl ester (38) (254 mg) as a gum (Found: M⁺, 530.1176. C₂₃H₃₁O₆ requires M, 530.1167); δ 0.96 (3 H, t, J 7.1 Hz, 2-CH₂CH₂Me), 1.19 (3 H, s, 18-H₃), 2.90 (1 H, d, J 10.0 Hz, 6-H; superimposed on signal at 2.95 p.p.m.), 3.75 (4 H, s, OMe and 3 α -H), 3.79 (1 H, d, J 10.3 Hz, 5-H), 4.35 (1 H, d, J 3.9 Hz, 1-H), 5.00 (1 H, br s, 17-H), and 5.28 (1 H, br s, 17-H); m/z 530 (M⁺, 73%) 471 (100), 403 (25), 385 (82), 371 (39), 357 (95), 343 (22), 339 (20), 325 (51), 297 (49), 281 (65), 279 (30), 105m (19), and 91 (36).

(g) ent-3 α ,10 β ,13-Trihydroxy-1 α -iodo-2 β -propyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (37). The 2 α -propyl-1(10),16-diene-7,19-dioic acid 7-methyl ester (29) (220 mg) in tetrahydrofuran (20 ml) and dichloromethane (40 ml) was stirred vigorously with saturated aqueous sodium hydrogen carbonate (40 ml) and iodine (155 mg) for 1 h. Work-up as in (a) gave 1 β -iodo-2 α -propyl-GA₁ methyl ester (37) (294 mg) as a gum (Found: M⁺, 530.1143. C₂₃H₃₁O₆ requires M, 530.1167); δ 0.90 (3 H, t, J 7.4 Hz, 2-CH₂CH₂Me), 1.21 (3 H, s, 18-H₃), 2.68 (1 H, d, J 9.5 Hz, 6-H; superimposed on signal at 2.6 p.p.m.), 3.75 (4 H, br s, OMe and 3 α -H), 3.83 (1 H, d, J 9.5 Hz, 5-H), 4.29 (1 H, s, 1-H), 5.00 (1 H, br s, 17-H), and 5.27 (1 H, br s, 17-H); m/z 530 (M⁺, 37%) 499 (45), 471 (65), 403 (37), 385 (80), 371 (67), 357 (100), 353 (33), 343 (61), 339 (35), 325 (78), 297 (61), 281 (96), 128 (67), 105 (41), and 91 (65).

Reduction of Iodo Lactones with Tri-butylstannane and 2,2'-Azobisisobutyronitrile (AIBN).—(a) ent-3 α ,10 β -Dihydroxy-2 α -methyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (42). 1 β -Iodo-2 β -methyl-GA₄ methyl ester (33) (76 mg) was refluxed in toluene (30 ml) with AIBN (catalytic) and tributylstannane, prepared from tributyltin chloride (222 μ l). After 1 h, evaporation under reduced pressure followed by flash chromatography with ethyl acetate-light petroleum (30:70) gave 2 β -methyl-GA₄ methyl ester (42) (56 mg), m.p. 135–136 °C (from ethyl acetate-light petroleum) [lit.⁴ m.p. 154–155 °C (from acetone-light petroleum)] (Found: M⁺ 360.1932. Calc. for C₂₁H₂₈O₅: M, 360.1936); identical by n.m.r. and mass spectrometry with the compound described previously.⁴

(b) ent-3 α ,10 β -Dihydroxy-2 β -methyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (41). Reduction of 1 β -iodo-2 α -methyl-GA₄ methyl ester (32) (50 mg), as in (a) under the same conditions and with the same quantities of reagents, with purification by flash chromatography [ethyl acetate-light petroleum (45:55)] gave 2 α -methyl-GA₄ methyl ester (41) (23 mg), m.p. 175–178 °C (from ethyl acetate-light petroleum) (Found: M⁺, 360.1968. C₂₁H₂₈O₅ requires M, 360.1936); δ 1.09 (3 H, d, J 8.1 Hz, 2 α -Me), 1.16 (3 H, s, 18-H₃), 2.65 (1 H, d, J 10.7 Hz, 6-H), 3.16 (1 H, d, J 10.7 Hz, 5-H), 3.56 (1 H, s, 3 α -H), 3.71 (3 H, s, OMe), 4.85 (1 H, br s, 17-H), and 4.98 (1 H, br s, 17-H); m/z 360 (M⁺, 5%), 342 (12), 328 (43), 300 (15), 298 (100), 282 (20), 238 (74), 223 (24), and 91 (28).

(c) ent-3 α ,10 β ,13-Trihydroxy-2 β -methyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (44). 1 β -Iodo-2 α -methyl-GA₁ methyl ester (34) (128 mg) was refluxed, as a suspension, in toluene (40 ml) containing AIBN (catalytic) and tributylstannane [from tributyltin chloride (550 μ l)]. After 1 h the mixture was evaporated under reduced pressure and the resultant gum was subjected to flash chromatography. After removal of tin residues by elution with ethyl acetate-light petroleum (5:95) further elution with ethyl acetate-light petroleum (60:40) gave 2 α -methyl-GA₁ methyl ester (44) (100 mg) as a gum (Found: M⁺, 376.1886. C₂₁H₂₈O₆ requires M,

376.1886); δ 1.08 (3 H, d, J 8.1 Hz, 2 α -Me), 1.15 (3 H, s, 18-H₃), 2.69 (1 H, d, J 10.3 Hz, 6-H; superimposed on 1 H, br m, 9-H), 3.17 (1 H, d, J 10.3 Hz, 5-H), 3.55 (1 H, br s, 3 α -H), 3.72 (3 H, s, OMe), 4.95 (1 H, br s, 17-H), and 5.25 (1 H, br s, 17-H); m/z 376 (M^+ , 29%), 361 (18), 359 (16), 358 (16), 345 (26), 344 (100), 317 (23), and 298 (21).

(d) ent-2 α -Ethyl-3 α ,10 β ,13-trihydroxy-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (**46**). 2 β -Ethyl-1 β -iodo-GA₁ methyl ester (**36**) (637 mg) in toluene (30 ml) was refluxed for 1 h with AIBN (catalytic) and tributylstannane (670 μ l). Flash chromatography of the product, first with light petroleum and ethyl acetate–light petroleum (1:9) to remove tin residues, and then with ethyl acetate–light petroleum (55:45), gave 2 β -ethyl-GA₁ methyl ester (**46**) as a gum (Found: M^+ , 390.2056; M^+ – 32, 358.1787. C₂₂H₃₀O₆ requires M , 390.2042; M – 32, 358.1780); δ 0.92 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.16 (3 H, s, 18-H₃), 2.68 (1 H, d, J 10.5 Hz, 6-H), 3.19 (1 H, d, J 10.5 Hz, 5-H), 3.72 (4 H, s, OMe and 3 α -H), 4.95 (1 H, br s, 17-H), and 5.26 (1 H, br s, 17-H); δ (C₅D₅N) 0.88 (3 H, t, J 7.1 Hz, 2-CH₂Me), 1.05 (3 H, s, 18-H₃), 3.01 (1 H, d, J 10.5 Hz, 6-H), 3.60 (3 H, s, OMe), 3.71 (1 H, d, J 10.5 Hz, 5-H), 3.96 (1 H, dd, J 5.4 and 3.2 Hz, 3 α -H), 5.08 (1 H, br s, 17-H), 5.63 (1 H, br s, 17-H), 6.46 (1 H, br s, 13-OH), and 7.08 (1 H, d, J 5.9 Hz, 3 β -OH) (on addition of D₂O the signals at 7.08 and 6.46 p.p.m. disappeared and the signal at 3.96 p.p.m. collapsed to 1 H, d, J 2.7 Hz); m/z 390 (M^+ , 69%), 372 (14), 358 (100), 344 (20), 333 (24), 331 (40), 330 (35), 313 (20), 312 (32), 304 (31), 267 (24), and 239 (23).

(e) ent-2 β -Ethyl-3 α ,10 β ,13-trihydroxy-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (**45**). 2 α -Ethyl-1 β -iodo-GA₁ methyl ester (**35**) (467 mg) in toluene (30 ml) was refluxed for 1 h with AIBN (catalytic) and tributylstannane (0.5 ml). Flash chromatography of the product as in (d) [except that the product was eluted with ethyl acetate–light petroleum (6:4)] gave 2 α -ethyl-GA₁ methyl ester (**45**) (239 mg) as a gum (Found: M^+ , 390.2074; M^+ – 32, 358.1757. C₂₂H₃₀O₆ requires M , 390.2042; M – 32, 358.1780); δ 0.91 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.53 (3 H, s, 18-H₃), 2.68 (1 H, d, J 10.0 Hz, 6-H), 3.19 (1 H, d, J 10.3 Hz, 5-H), 3.61 (1 H, s, 3 α -H), 3.72 (3 H, s, OMe), 4.95 (1 H, br s, 17-H), and 5.25 (1 H, br s, 17-H); m/z 390 (M^+ , 27%), 372 (13), 358 (100), 344 (11), 331 (20), 330 (13), 312 (21), 304 (18), 267 (10), and 239 (13); later column fractions gave slightly impure 2 α -ethyl-GA₁ methyl ester (**45**) (80 mg).

(f) ent-3 α ,10 β ,13-Trihydroxy-2 β -propyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (**47**). The product, obtained as in (a) from 1 β -iodo-2 α -propyl-GA₁ methyl ester (**37**) (180 mg) in toluene (30 ml; deoxygenated with nitrogen), AIBN (catalytic), and tributylstannane (0.3 ml), was subjected to flash chromatography. After elution of tin residues with ethyl acetate–light petroleum (5:95) elution with ethyl acetate–light petroleum (1:1) gave 2 α -propyl-GA₁ methyl ester (**47**) (73 mg, m.p. 203–205 °C (from ethyl acetate–light petroleum) (Found: C, 68.4; H, 7.9. C₂₃H₃₂O₆ requires C, 68.3; H, 8.0%); δ 0.87 (3 H, t, J 5.6 Hz, 2-CH₂CH₂Me), 1.16 (3 H, s, 18-H₃), 2.69 (1 H, d, J 10.0 Hz, 6-H), 3.19 (1 H, d, J 10.2 Hz, 5-H), 3.61 (1 H, s, 3 α -H), 3.72 (3 H, s, OMe), 4.95 (1 H, br s, 17-H), and 5.25 (1 H, br s, 17-H); m/z 404 (M^+ , 26%), 386 (15), 372 (100), 358 (19), 345 (26), 344 (22), 326 (33), 304 (30), 281 (12), 239 (37), 135 (37), 128 (33), 121 (33), 115 (30), 105 (37), and 91 (89).

(g) ent-3 α ,10 β ,13-Trihydroxy-2 α -propyl-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (**48**). The product, obtained as in (a) from 1 β -iodo-2 β -propyl-GA₁ methyl ester (**38**) (352 mg) in toluene (30 ml; deoxygenated with nitrogen), AIBN (catalytic), and tributylstannane (0.5 ml), was flash chromatographed. After elution of tin residues as in (f), elution with ethyl acetate–light petroleum (1:1) gave 2 β -propyl-GA₁ methyl ester (**48**) (261 mg) as a gum (Found: M^+ , 404.2186. C₂₃H₃₂O₆ requires M , 404.2199); δ 0.92 (3 H, t, J 6.4 Hz, 2-

CH₂CH₂Me), 1.15 (3 H, s, 18-H₃), 2.68 (1 H, d, J 10.2 Hz, 6-H), 3.18 (1 H, d, J 10.5 Hz, 5-H), 3.67 (1 H, superimposed on signal at 3.71, 3 α -H), 3.71 (3 H, s, OMe), 4.95 (1 H, br s, 17-H), and 5.25 (1 H, br s, 17-H); m/z 404 (M^+ , 81%), 373 (39), 372 (100), 345 (51), 344 (58), 327 (28), 326 (55), 304 (49), 281 (49), 239 (62), 135 (65), 121 (46), 105 (47), and 91 (89).

(h) ent-3 α -Acetoxy-2 β -ethyl-10 β -hydroxy-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (**53**). To 2 α -ethyl-GA₁ 3-acetate 13-methyl oxalate 7-methyl ester (**51**) (158 mg) in refluxing toluene (20 ml) were added tributylstannane (170 μ l) and AIBN (catalytic). The reaction was monitored by t.l.c. After 0.5 h more tributylstannane (250 μ l) and AIBN (catalytic) were added, and after a further 15 min AIBN (catalytic) was again added. After 0.5 h (total reflux time 1.25 h) the reaction was complete. Tin residues were removed from the crude product by gravity elution through silica gel (60–120 mesh) with light petroleum and then ethyl acetate–light petroleum (5:95). Elution with ethyl acetate–light petroleum (1:1) gave 2 α -ethyl-GA₄ 3-acetate methyl ether (**53**) (74 mg) as a gum (Found: M^+ – 31, 385.2014; M^+ – 60, 356.1974. C₂₄H₃₂O₆ requires M – 31, 385.2015; M – 60, 356.1987); δ 0.87 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.03 (3 H, s, 18-H₃), 2.17 (3 H, s, COMe), 2.63 (1 H, br t, 9- or 13-H; under signal at 2.67 p.p.m.), 2.67 (1 H, d, J 11.0 Hz, 6-H; superimposed on signal at 2.63 p.p.m.), 3.11 (1 H, d, J 11.0 Hz, 5-H), 3.70 (3 H, s, OMe), 4.86 (1 H, br s, 17-H), 4.99 (1 H, br s, 17-H), and 5.16 (1 H, d, J 3.9 Hz, 3 α -H); m/z 385 (M^+ – 31, 6%), 356 (15), 312 (100), 296 (22), 283 (23), 253 (39), 252 (44), 223 (90), and 43 (38).

(i) ent-3 α -Acetoxy-2 α -ethyl-10 β -hydroxy-20-norgibberell-16-ene-7,19-dioic acid 7-methyl ester 19,10-lactone (**54**). To 2 β -ethyl-GA₁ 3-acetate 13-methyl oxalate 7-methyl ester (**52**) (214 mg) in refluxing toluene (20 ml) were added tributylstannane (300 μ l) and AIBN (catalytic). After 1 h tin residues were removed from the recovered product by gravity feed elution through silica gel (60–120 mesh) with light petroleum and ethyl acetate–light petroleum (5:95). Elution with ethyl acetate–light petroleum (1:1) yielded product (127 mg), which was further purified by flash chromatography with ethyl acetate–light petroleum (30:70) to give 2 β -ethyl-GA₄ 3-acetate methyl ether (**54**) (71 mg) as a gum (Found: M^+ , 416.2230. C₂₄H₃₂O₆ requires M , 416.2199); δ 0.87 (3 H, t, J 7.3 Hz, 2-CH₂Me), 1.03 (3 H, s, 18-H₃), 2.17 (3 H, s, COMe), 2.63 (1 H, br t, 9- or 13-H; under signal at 2.67 p.p.m.), 2.67 (1 H, d, J 11.0 Hz, 6-H; superimposed on signal at 2.63 p.p.m.), 3.11 (1 H, d, J 11.0 Hz, 5-H), 3.70 (3 H, s, OMe), 4.86 (1 H, br s, 17-H), 4.99 (1 H, br s, 17-H), and 5.16 (1 H, d, J 3.9 Hz, 3 α -H); m/z 416 (M^+ , 3%), 385 (10), 356 (34), 324 (48), 312 (100), 296 (49), 253 (37), 252 (41), 251 (29), 223 (85), and 43 (46).

Preparation of 2 α -Methylgibberellin A₄ Methyl Ester (41) from 2-Methylgibberellin A₇ Methyl Ester (16).—(a) *Epoxidation.* 2-Methylgibberellin A₇ methyl ester (**16**) (164 mg), was stirred with *m*-chloroperbenzoic acid (94 mg) in chloroform (10 ml) for 16 h at 5 °C. Dilution with chloroform (10 ml) was followed by washing with saturated aqueous sodium hydrogen carbonate (15 ml) then water (15 ml). After drying (MgSO₄), evaporation under reduced pressure followed by flash chromatography with ethyl acetate–light petroleum (1:1) yielded 2-methyl-GA₇ 16 α ,17-epoxide methyl ester (99 mg) as a gum; δ 1.24 (3 H, s, 18-H₃), 1.82 (3 H, d, J 1.5 Hz, 2-Me), 2.81 (d, J 11 Hz, 6-H; one arm under signal at 2.83 p.p.m.), 2.83 (2 H, s, 17-H₂), 3.73 (3 H, s, OMe), 3.93 (1 H, br s, 3 α -H), and 6.03 (1 H, d, J 1.7 Hz, 1-H); earlier fractions (9 mg) contained a slightly less polar compound by t.l.c., presumably the 16 β ,17-epoxide.

(b) *Hydrogenation.* 2-Methyl-GA₇ 16 α ,17-epoxide methyl ester (70 mg) in tetrahydrofuran (8 ml) containing 10% palladium–calcium carbonate (20 mg) was stirred in hydrogen at ambient temperature and pressure. After 1 h, filtration

through Celite and evaporation under reduced pressure gave a gum which lacked olefinic signals in its ^1H n.m.r. spectrum.

(c) *De-epoxidation*. To a solution of sodium iodide (400 mg) and sodium acetate (150 mg) in glacial acetic acid (7 ml), acetone (1 ml), and water (0.5 ml) was added zinc powder (400 mg; freshly activated with 2M hydrochloric acid), followed dropwise by a solution of the hydrogenation product in acetone (3 ml). After 3.5 h the mixture was diluted with ethyl acetate (20 ml) and washed with saturated aqueous sodium hydrogen carbonate (20 ml). After drying (MgSO_4), evaporation under reduced pressure gave a gum (16 mg). Although slightly impure, the major product was identified by t.l.c. and ^1H n.m.r. as 2 α -methyl- GA_4 methyl ester (**41**), described earlier.

ent-3 α -Acetoxy-2 α -ethyl-10 β ,13-dihydroxy-20-norgibberell-16-ene-7,19-dioic Acid 7-Methyl Ester 19,10-Lactone (50).—2 β -Ethyl- GA_1 methyl ester (**46**) (317 mg) in pyridine (5 ml) was stirred with redistilled acetic anhydride (2 ml) for 10 h. Azeotropic removal of pyridine and acetic anhydride with toluene under reduced pressure gave 2 β -ethyl- GA_1 3-acetate methyl ester (**50**) (276 mg) as a gum (Found: M^+ , 432.2164. $\text{C}_{24}\text{H}_{32}\text{O}_7$ requires M , 432.2148); δ 0.87 (3 H, t, J 7.3 Hz, 2- CH_2Me), 1.03 (3 H, s, 18- H_3), 2.16 (3 H, s, COMe), 2.66 (1 H, d, J 10.5 Hz, 6-H), 3.12 (1 H, d, J 10.5 Hz, 5-H), 3.72 (3 H, s, OMe), 4.96 (1 H, br s, 17-H), 5.15 (1 H, d, J 3.9 Hz, 3 α -H), and 5.27 (1 H, br s, 17-H); m/z 432 (M^+ , 13%), 414 (6), 401 (17), 400 (19), 372 (85), 354 (17), 340 (62), 328 (100), 312 (58), 304 (35), 267 (41), 239 (68), and 43 (64). The polarity of the product by t.l.c. was identical with that of starting material.

ent-3 α -Acetoxy-2 β -ethyl-10 β ,13-dihydroxy-20-norgibberell-16-ene-7,19-dioic Acid 7-Methyl Ester 19,10-Lactone (49).—2 α -Ethyl- GA_1 methyl ester (**45**) (179 mg) in pyridine (5 ml) was stirred with redistilled acetic anhydride (1 ml) for 5 h. Work-up as in the previous experiment gave 2 α -ethyl- GA_1 3-acetate methyl ester (**49**) (199 mg) as a gum (Found: M^+ , 432.2148. $\text{C}_{24}\text{H}_{32}\text{O}_7$ requires M , 432.2139); δ 0.94 (3 H, t, J 7.3 Hz, 2- CH_2Me), 1.05 (3 H, s, 18- H_3), 2.13 (3 H, s, COMe), 2.68 (1 H, d, J 10.3 Hz, 6-H), 3.18 (1 H, d, J 10.3 Hz, 5-H), 3.73 (3 H, s, OMe), 4.83 (1 H, s, 3 α -H), 4.96 (1 H, br s, 17-H), and 5.27 (1 H, br s, 17-H); m/z 432 (M^+ , 9%), 401 (20), 400 (38), 372 (76), 340 (34), 328 (100), 313 (23), 312 (27), 304 (21), 299 (20), 269 (27), 268 (25), 267 (20), 239 (49), and 432 (41). (N.B. T.l.c. polarity of product was identical with that of starting material.)

ent-3 α -Acetoxy-2 α -ethyl-10 β -hydroxy-13-(2-methoxy-2,3-dioxoethoxy)-20-norgibberell-16-ene-7,19-dioic Acid 7-Methyl Ester 19,10-Lactone (52).—2 β -Ethyl- GA_1 3-acetate methyl ester (**50**) (276 mg) in dry tetrahydrofuran (25 ml) was refluxed with oxalyl chloride (2 ml) for 1 h. The mixture was cooled in an ice-bath and water (15 ml) was added cautiously down the condenser. Extraction with ethyl acetate (3 \times 30 ml) gave, after recovery, a brown gum. The crude product was methylated with ethereal diazomethane and then subjected to flash chromatography with ethyl acetate–light petroleum (40:60), which yielded the oxalyl ester (**52**) (214 mg) as a gum (pure by t.l.c. but ^1H n.m.r. shows a non- GA impurity, possibly dimethyl oxalate) (Found: M^+ – 31, 487.1931. $\text{C}_{27}\text{H}_{34}\text{O}_{10}$ requires M – 31, 487.1968); δ 0.87 (3 H, t, J 7.3 Hz, 2- CH_2Me), 1.03 (3 H, s, 18- H_3), 2.17 (3 H, s, COMe), 2.69 (1 H, d, J 10.8 Hz, 6-H), 3.15 (1 H, d, J 10.8 Hz, 5-H), 3.73 (3 H, s, OMe), 3.89 (3 H, s, 13- OCOCO_2Me), 5.07 (1 H, br s, 17-H), 5.15 (1 H, d, J 3.7 Hz, 3 α -H), and 5.26 (1 H, br s, 17-H) (the impurity gave rise to two singlets at 3.973 and 3.966 p.p.m.); m/z 487 (M^+ – 31, 13%), 458 (68), 426 (40), 414 (100), 398 (57), 355 (63), 354 (80), 325 (79), 311 (63), 310 (52), 251 (43), 211 (56), 59 (48), and 43 (87).

ent-3 α -Acetoxy-2 β -ethyl-10 β -hydroxy-13-(2-methoxy-2,3-dioxoethoxy)-20-norgibberell-16-ene-7,19-dioic Acid 7-Methyl Ester 19,10-Lactone (51).—2 α -Ethyl- GA_1 3-acetate methyl ester (**49**) (199 mg) in dry tetrahydrofuran (15 ml) was refluxed with oxalyl chloride (2 ml) for 1 h. Quenching and work-up as for the 2 β -ethyl isomer, as described in the previous experiment, gave crude product which was methylated with ethereal diazomethane. Flash chromatography with ethyl acetate–light petroleum (1:1) yielded the oxalyl ester (**51**) (158 mg) as a gum (Found: M^+ – 31, 487.1943. $\text{C}_{27}\text{H}_{34}\text{O}_{10}$ requires M – 31, 487.1968); δ 0.94 (3 H, t, J 7.3 Hz, 2- CH_2Me), 1.05 (3 H, s, 18- H_3), 2.13 (3 H, s, COMe), 2.71 (1 H, d, J 10.7 Hz, 6-H), 3.20 (1 H, d, J 11.1 Hz, 5-H), 3.74 (3 H, s, Me), 3.89 (3 H, s, 13- OCOCO_2Me), 4.83 (1 H, s, 3 α -H), 5.07 (1 H, br s, 17-H), and 5.25 (1 H, br s, 17-H); m/z 487 (M^+ – 31, 12%), 458 (73), 426 (21), 415 (30), 414 (100), 355 (54), 354 (63), 325 (79), 311 (64), 310 (51), 251 (37), 221 (42), 91 (20), 59 (24), and 43 (45).

ent-2 α -Ethyl-3 α ,10 β -dihydroxy-20-norgibberell-16-ene-7,19-dioic Acid 7-Methyl Ester 19,10-Lactone (56).—2 β -Ethyl- GA_4 3-acetate methyl ester (**54**) (70 mg) was stirred, in methanol (7 ml), with saturated aqueous potassium carbonate (7 ml), for 2 h. The mixture was diluted with water (10 ml), acidified to pH 2, and extracted with ethyl acetate (3 \times 20 ml). Flash chromatography of the recovered gum with ethyl acetate–light petroleum (30:70) gave, sequentially, starting material (**54**) (6 mg) and 2 β -ethyl- GA_4 methyl ester (**56**) (29 mg) as a gum (Found: M^+ , 374.2011; M^+ – 18, 356.2024; M^+ – 32, 342.2796. $\text{C}_{22}\text{H}_{30}\text{O}_5$ requires M , 374.2095; M – 18, 356.1987; M – 32, 342.1831); δ 0.92 (3 H, t, J 7.3 Hz, 2- CH_2Me), 1.16 (3 H, s, 18- H_3), 2.63 (1 H, br t, 9- or 13-H), 2.69 (1 H, d, J 11.0 Hz, 6-H), 3.16 (1 H, d, J 10.7 Hz, 5-H), 3.70 (4 H, s, OMe and 3 α -H), 4.85 (1 H, br s, 17-H), and 4.97 (1 H, br s, 17-H); δ ($\text{C}_5\text{D}_5\text{N}$) 0.87 (3 H, t, J 7.3 Hz, 2- CH_2Me), 1.51 (3 H, s, 18- H_3), 2.54 (1 H, br t, 9- or 13-H), 2.99 (1 H, d, J 11.0 Hz, 6-H), 3.66 (3 H, s, OMe), 3.93 (1 H, d, 5-H; partly obscured by the signal at 3.66 p.p.m.), 3.96 (1 H, br dd, J ca. 5.9 and 2.2 Hz, 3 α -H), 4.91 (1 H, br s, 17-H), 5.01 (1 H, br s, 17-H), and 7.10 (1 H, d, J 5.9 Hz, 3 β -OH) (on addition of D_2O the signal at 3.96 p.p.m. collapsed to a br d, J 2.2 Hz); m/z 374 (M^+ , 1%), 356 (9), 342 (39), 312 (70), 296 (33), 252 (100), 223 (79), 91 (46), and 41 (47).

ent-2 β -Ethyl-3 α ,10 β -dihydroxy-20-norgibberell-16-ene-7,19-dioic Acid 7-Methyl Ester 19,10-Lactone (55).—To 2 α -ethyl- GA_4 3-acetate methyl ester (**53**) (72 mg) in methanol (20 ml) was added saturated aqueous potassium carbonate (to pH 9–10). After stirring for 2.5 h, t.l.c. indicated ca. 70% conversion into a more polar product. Saturated aqueous potassium carbonate (6 ml) was added and the mixture stirred for 2 min. After acidification to pH 2 the mixture was extracted with ethyl acetate (3 \times 20 ml). Flash chromatography of the recovered gum with ethyl acetate–light petroleum (30:70) gave, sequentially, a mixture (27 mg) of starting material (**53**) and 2 α -ethyl- GA_4 methyl ester (**55**), and pure 2 α -ethyl- GA_4 methyl ester (**55**) (43 mg) as a gum (Found: M^+ , 374.2110. $\text{C}_{22}\text{H}_{30}\text{O}_5$ requires M , 374.2093); δ 0.91 (3 H, t, J 7.3 Hz, 2- CH_2Me), 1.16 (3 H, s, 18- H_3), 2.63 (1 H, br t, 9- or 13-H), 2.70 (1 H, d, J 10.7 Hz, 6-H), 3.18 (1 H, d, J 10.7 Hz, 5-H), 3.63 (1 H, s, 3-H), 3.71 (3 H, s, OMe), 4.85 (1 H, br s, 17-H), and 4.98 (1 H, br s, 17-H); m/z 374 (M^+ , 4%), 356 (21), 343 (25), 342 (51), 328 (23), 312 (100), 253 (34), 252 (67), 223 (76), and 91 (30).

Hydrolyses of Methyl Esters with Sodium Propane-1-thiolate.—(a) *ent-3 α ,10 β -Dihydroxy-2 α -methyl-20-norgibberell-16-ene-7,19-dioic acid 19,10-lactone (4)*. 2 β -Methyl- GA_4 methyl ester (**42**) (50 mg) was stirred with sodium propane-1-thiolate solution (1.5 ml) for 5 h under nitrogen. After work-up the ethyl acetate extracts were washed with saturated aqueous sodium

hydrogen carbonate (3 × 20 ml). The aqueous washings were acidified to pH 2 and extracted with ethyl acetate (3 × 20 ml). The organic extracts were dried (MgSO₄); evaporation under reduced pressure followed by flash chromatography with ethyl acetate–light petroleum–acetic acid (40:60:1) yielded 2β-methyl-GA₄ (**4**) (28 mg) as a gum (Found: *M*⁺, 346.1760. Calc. for C₂₀H₂₆O₅: *M*, 346.1773); identical (n.m.r. and mass spectrometry) with the previously prepared sample.² G.l.c.–mass spectrometric analysis indicated the presence of 6% *endo*-isomer.

(b) *ent*-3α,10β-Dihydroxy-2β-methyl-20-norgibberell-16-ene-7,19-dioic acid 19,10-lactone (**3**). 2α-Methyl-GA₄ methyl ester (**41**) (22 mg) was demethylated and the product worked up as described for the 2β-isomer (**42**) in the previous experiment. Purification by flash chromatography ethyl acetate–light petroleum–acetic acid (40:60:1) gave 2α-methyl-GA₄ (**3**) (20 mg) as a gum (Found: *M*⁺, 346.1776. C₂₀H₂₆O₅ requires *M*, 346.1773); δ[(CD₃)₂CO] 1.03 (3 H, d, *J* 8.1 Hz, 2α-Me), 1.13 (3 H, s, 18-H₃), 2.60 (1 H, d, *J* 10.5 Hz, 6-H), 3.18 (1 H, d, *J* 10.7 Hz, 5-H), 3.45 (1 H, s, 3α-H), 4.84 (1 H, br s, 17-H), and 4.97 (1 H, br s, 17-H); g.l.c.–mass spectrum of the Me ester *O*-SiMe₃ derivative: *m/z* 432 (*M*⁺, 38%), 342 (51), 310 (21), 298 (100), 289 (67), 261 (32), 239 (50), 238 (57), 233 (56), 201 (29), and 143 (71). G.l.c.–mass spectrometry indicated the presence of 5% *endo*-isomer.

(c) *ent*-2β-Ethyl-3α,10β-dihydroxy-20-norgibberell-16-ene-7,19-dioic acid 19,10-lactone (**5**). 2α-Ethyl-GA₄ methyl ester (**55**) (43 mg) was stirred with sodium propane-1-thiolate solution (1.5 ml) under nitrogen for 5 h. Work-up followed by flash chromatography [ethyl acetate–light petroleum–acetic acid (30:70:1)] gave 2α-ethyl-GA₄ (**5**) (30 mg) as a gum (Found: *M*⁺, 360.1951. C₂₂H₃₀O₅ requires *M*, 360.1936); δ[(CD₃)₂CO] 0.89 (3 H, t, *J* 7.3 Hz, 2-CH₂Me), 1.13 (3 H, s, 18-H₃), 2.59 (1 H, d, *J* 10.7 Hz, 6-H; superimposed on signal at 2.61 p.p.m.), 2.61 (1 H, br t, 9- or 13-H; under signal at 2.59 p.p.m.), 3.20 (1 H, d, *J* 10.7 Hz, 5-H), 3.54 (1 H, s, 3α-H), 4.84 (1 H, br s, 17-H), and 4.97 (1 H, br s, 17-H); *m/z* 360 (*M*⁺, 6%), 342 (19), 324 (16), 314 (43), 298 (100), 296 (43), 269 (59), 223 (65), and 91 (31); g.l.c.–mass spectrometry of the Me ester *O*-SiMe₃ derivative: *m/z* 446 (*M*⁺, 45%), 356 (59), 312 (94), 289 (78), 261 (40), 253 (44), 252 (40), 233 (57), 223 (48), and 157 (100).

(d) *ent*-2α-Ethyl-3α,10β-dihydroxy-20-norgibberell-16-ene-7,19-dioic acid 19,10-lactone (**6**). 2α-Ethyl-GA₄ methyl ester (**56**) (28 mg) was stirred for 6 h with sodium propane-1-thiolate solution (1.8 ml) as in (a). Flash chromatography of the crude acidic product with ethyl acetate–light petroleum–acetic acid (35:65:1) gave 2β-ethyl-GA₄ (**6**) (23 mg) as a gum (Found: *M*⁺, 360.1927. C₂₁H₂₈O₅ requires *M*, 360.1936); δ[(CD₃)₂CO] 0.91 (3 H, t, *J* 7.3 Hz, 2-CH₂Me), 1.13 (3 H, s, 18-H₃), 2.59 (1 H, d, *J* 10.7 Hz, 6-H; superimposed on signal at 2.62 p.p.m.), 2.62 (1 H, br t, 9- or 13-H; under signal at 2.59 p.p.m.), 3.16 (1 H, d, *J* 10.7 Hz, 5-H), 3.62 (1 H, d, *J* 2.9 Hz, 3α-H), 4.85 (1 H, br s, 17-H), and 4.97 (1 H, br s, 17-H); *m/z* 360 (*M*⁺, 3%), 342 (28), 324 (34), 314 (38), 298 (100), 296 (38), 223 (54), and 91 (35); g.l.c.–mass spectrometry of the Me ester *O*-SiMe₃ derivative: *m/z* 446 (*M*⁺, 71%), 418 (17), 356 (55), 312 (45), 289 (84), 261 (39), 253 (39), 233 (52), and 157 (100).

(3) *ent*-3α,10β,13-Trihydroxy-2β-methyl-20-norgibberell-16-ene-7,19-dioic acid 19,10-lactone (**7**). 2α-Methyl-GA₁ methyl ester (**44**) (99 mg) was stirred with sodium propane-1-thiolate solution (1.8 ml) for 6.5 h and the product worked up as described in (a). Flash chromatography with ethyl acetate–light petroleum (60:40) gave 2α-methyl-GA₁ (**7**) (20 mg) as a gum (Found: *M*⁺, 362.1703. C₂₀H₂₆O₆ requires *M*, 362.1729); δ[(CD₃)₂CO] 1.03 (3 H, d, *J* 8.1 Hz, 2α-Me), 1.13 (3 H, s, 18-H₃), 2.58 (1 H, d, *J* 10.3 Hz, 6-H), 3.19 (1 H, d, *J* 10.3 Hz, 5-H), 3.45 (1 H, s, 3α-H), 4.86 (1 H, br s, 17-H), and 5.19 (1 H, br s, 17-H); *m/z* 362 (*M*⁺, 31%), 344 (86), 316 (25), 298 (51), 277 (44), 230 (85),

228 (62), 154 (62), 152 (55), 91 (49), 77 (59), 55 (53), and 28 (100); g.l.c.–mass spectrum of the Me ester *O*-SiMe₃ derivative: *m/z* 520 (*M*⁺, 100%), 448 (13), 377 (15), 207 (25), and 73 (22); earlier and later column fractions yielded slightly impure 2α-methyl-GA₁ (**7**) (41 mg).

(f) *ent*-2β-Ethyl-3α,10β,13-trihydroxy-20-norgibberell-16-ene-7,19-dioic acid 19,10-lactone (**9**). The crude acid, obtained as in (a) from 2α-ethyl-GA₁ methyl ester (**45**) (94 mg) and sodium propane-1-thiolate solution (1.4 ml) was subjected to flash chromatography. Elution with ethyl acetate–light petroleum–acetic acid (40:60:1) gave 2α-ethyl-GA₁ (**9**) (23 mg) as a gum (Found: *M*⁺, 376.1861. C₂₁H₂₈O₅ requires *M*, 376.1886); δ[(CD₃)₂CO] 0.89 (3 H, t, *J* 7.3 Hz, 2-CH₂Me), 1.13 (3 H, s, 18-H₃), 2.28 (1 H, d, 6-H); partially obscured by other signals), 2.60 (1 H, br t, 9-H), 3.21 (1 H, d, *J* 10.3 Hz, 5-H), 3.54 (1 H, s, 3α-H), 4.87 (1 H, br s, 17-H), and 5.19 (1 H, br s, 17-H); *m/z* 390 (*M*⁺, 27%), 372 (13), 359 (29), 358 (100), 344 (11), 331 (20), 330 (13), 312 (21), 304 (18), 239 (13), and 163 (18); g.l.c.–mass spectrometry of the Me ester *O*-SiMe₃ derivative: *m/z* 534 (*M*⁺, 100%), 519 (6), 475 (5), 448 (12), 377 (15), 207 (19), and 73 (22); later column fractions yielded slightly impure 2α-ethyl-GA₁ (**9**) (37 mg).

(g) *ent*-3α,10β,13-Trihydroxy-2β-propyl-20-norgibberell-16-ene-7,19-dioic acid 19,10-lactone (**11**). 2α-Propyl-GA₁ methyl ester (**47**) (50 mg) was stirred with sodium propane-1-thiolate solution (1.5 ml) for 4 h as in (a). Flash chromatography of the crude product with ethyl acetate–light petroleum–acetic acid (35:65:2) gave 2α-propyl-GA₁ (**11**) (35 mg) as a solid foam (Found: *M*⁺, 390.2061; *M*⁺ – 18, 372.1942. C₂₂H₃₀O₆ requires *M*⁺, 390.2042; *M*⁺ – 18, 372.1937); δ[(CD₃)₂CO] 0.88 (3 H, t, *J* 7.0 Hz, 2-CH₂CH₂Me), 1.13 (3 H, br s, 18-H₃), 2.75 (1 H, d, *J* 10.3 Hz, 6-H), 3.19 (1 H, d, *J* 10.5 Hz, 5-H), 3.60 (1 H, d, *J* 3.4 Hz, 3α-H), 4.87 (1 H, br s, 17-H), and 5.20 (1 H, br s, 17-H); *m/z* 390 (*M*⁺, 63%), 372 (100), 345 (25), 344 (40), 328 (17), 326 (16), 326 (33), 290 (55), 281 (15), 239 (20), 163 (24), 136 (21), 135 (37), 129 (19), 128 (31), 127 (19), 121 (22), 120 (34) 105 (22), and 91 (36).

(h) *ent*-3α,10β,13-Trihydroxy-2α-propyl-20-norgibberell-16-ene-7,19-dioic acid 19,10-lactone (**12**). 2β-Propyl-GA₁ methyl ester (**48**) (49 mg) was stirred with sodium propane-1-thiolate solution (1.5 ml) for 6 h as in (a). Flash chromatography of the crude product with ethyl acetate–light petroleum–acetic acid (35:65:2) gave 2β-propyl-GA₁ (**12**) (38 mg) as a solid foam (Found: *M*⁺, 390.2047; *M*⁺ – 18, 372.1927. C₂₂H₃₀O₆ requires *M*, 390.2042; *M*⁺ – 18, 372.1937); δ[(CD₃)₂CO] 0.88 (3 H, t, *J* 6.9 Hz, 2-CH₂CH₂Me), 1.13 (1 H, s, 18-H₃), 2.58 (1 H, d, *J* 10.5 Hz, 6-H), 3.18 (1 H, d, *J* 10.2 Hz, 5-H), 3.60 (1 H, d, *J* 3.4 Hz, 3α-H), 4.87 (1 H, br s, 17-H), and 5.20 (1 H, br s, 17-H); *m/z* 390 (*M*⁺, 57%), 372 (100), 345 (19), 344 (32), 327 (15), 326 (30), 290 (54), 281 (13), 239 (20), 163 (25), 136 (17), 135 (26), 129 (12), 128 (13), 122 (15), 121 (20), 105 (17), and 91 (27).

(i) *ent*-2β,3α-Dihydroxy-2α-methyl-20-norgibberella-(10),16-diene-7,19-dioic acid 19,2-lactone (**60**). 2-Methyl-GA₇ methyl ester (**16**) (114 mg) was stirred for 5 h with sodium propane-1-thiolate solution (3 ml) under nitrogen. Work-up followed by flash chromatography with ethyl acetate–light petroleum–acetic acid (40:60:1) gave the 2β-methyl 19,2-lactone (**60**) (91 mg) as a gum (Found: *M*⁺, 344.1635. C₂₀H₂₄O₅ requires *M*, 344.1624); δ 1.25 (3 H, s, 18-H₃), 1.59 (3 H, s, 2β-Me), 3.23 (1 H, dd, *J* 5.8 and 2.2 Hz, 5-H), 3.92 (1 H, s, 3α-H), 4.92 (2 H, br s, 17-H₂), and 5.52 (1 H, br t, *J* 2 Hz × 2, 1-H); δ[(CD₃)₂CO] 1.16 (3 H, s, 18-H₃), 1.51 (3 H, s, 2β-Me), 2.42 (1 H, d, *J* 6.4 Hz, 6-H), 3.28 (1 H, dd, *J* 6.4 and 2.4 Hz, 5-H), 3.94 (1 H, s, 3α-H), 4.91 (2 H, br s, 17-H₂), and 5.54 (1 H, br s, 1-H); *m/z* 344 (*M*⁺, 22%), 326 (29), 299 (26), 298 (50), 282 (67), 281 (75), 253 (34), 237 (100), 193 (29), 167 (58), 93 (30), and 91 (57); g.l.c.–mass spectrum (of the Me ester *O*-SiMe₃ derivative): 430 (*M*⁺, 9%), 398 (34), 370 (25), 312 (36),

296 (41), 295 (38), 283 (45), 256 (35), 237 (92), 236 (100), and 75 (57).

Treatment of Iodo Lactones with 1,8-Diazabicyclo[5.4.0]undec-7-ene (DBU).—(a) 1β -Iodo- 2β -methylgibberellin A_4 methyl ester (**33**). The iodo lactone (**33**) (999 mg) was refluxed in toluene (40 ml) containing DBU (615 μ l). After 35 min the mixture was cooled, washed with aqueous hydrochloric acid (pH 3; $\times 2$) and the aqueous washings were back-extracted with ethyl acetate. The organic fractions were combined, dried ($MgSO_4$), and evaporated under reduced pressure. Flash chromatography with ethyl acetate–light petroleum (25:75) gave 2-methyl- GA_7 methyl ester (**16**) as prisms (571 mg), m.p. 168–172 °C [from ethyl acetate–acetone (trace)–light petroleum] (Found: C, 70.6; H, 7.5. $C_{21}H_{26}O_5$ requires C, 70.4; H, 7.3%); δ 1.24 (3 H, s, 18- H_3), 1.81 (3 H, d, J 1.0 Hz, 2-Me), 2.77 (1 H, d, J 11.0 Hz, 6-H), 3.17 (1 H, d, J 11.0 Hz, 5-H), 3.72 (3 H, s, OMe), 3.92 (1 H, br s, 3 α -H), 4.85 (1 H, br s, 17-H), 4.99 (1 H, br s, 17-H), and 6.01 (1 H, q, J 1.5 Hz, 1-H); m/z 358 (M^+ , 6%), 326 (21), 298 (17), 295 (100), 253 (30), 237 (74), 236 (55), 235 (42), 135 (18), and 91 (27).

(b) 1β -Iodo- 2α -methylgibberellin A_1 methyl ester (**34**). The iodo lactone (**34**) (31 mg) was refluxed in toluene (20 ml) containing DBU (50 μ l) for 1 h. After cooling, the mixture was evaporated under reduced pressure. Purification by flash chromatography with ethyl acetate–light petroleum (1:1) afforded starting material (**34**) (identified by 1H n.m.r. and t.l.c.).

(c) 1β -Iodo- 2β -methylgibberellin A_1 (**39**). The iodo lactone (**39**) (58 mg) was converted into the tributylstannyl ester by refluxing in toluene (15 ml) with bis(tributylstannyl) oxide (30 μ l) and removing the water with a Dean–Stark trap. After 30 min, DBU (36 μ l) was added and refluxing was continued for a further 30 min. Evaporation under reduced pressure gave a gum which was purified by flash chromatography with ethyl acetate–light petroleum–acetic acid (75:25:1) to give ent- 2β - 3α , 2β ,13-trihydroxy- 2α -methyl-20-norgibberell-1(10),16-diene-7,19-dioic acid 19,2-lactone (**60**) (ca. 45 mg) as a gum (Found: M^+ , 360.1580. $C_{20}H_{24}O_6$ requires M , 360.1573); δ 1.16 (3 H, s, 18- H_3), 1.51 (3 H, s, 2β -Me), 2.43 (1 H, d, J 5.9 Hz, 6-H), 3.27 (1 H, dd, J 6.0 and 2.5 Hz, 5-H), 3.95 (1 H, s, 3 α -H), 4.92 (1 H, br s, 17-H), 5.09 (1 H, br s, 17-H), and 5.55 (1 H, br s, 1-H); m/z 360 (M^+ , 8%), 342 (26), 314 (20), 299 (19), 298 (84), 297 (15), 266 (18), 253 (49), 169 (28), 44 (100), and 43 (78); g.l.c.–mass spectrometry of the Me ester O -SiMe $_3$ derivative: m/z 518 (M^+ , 100%), 503 (6), 489 (12), 384 (21), 383 (25), 369 (10), and 238 (19).

The same result was obtained when 1β -iodo- 2β -methyl- GA_1 (**39**) (58 mg) was refluxed for 1 h in tetrahydrofuran (15 ml) containing DBU (36 μ l).

(d) 2β -Ethyl- 1β -iodogibberellin A_1 methyl ester (**36**). The iodo lactone (**36**) (140 mg) was refluxed for 1 h in toluene (20 ml) containing DBU (100 μ l). Work-up as in (a) followed by flash chromatography with ethyl acetate–light petroleum (60:40) gave, sequentially, as gums, ent- 2α -ethyl- 3α , 2β ,13-trihydroxy-20-norgibberella-1(10),16-diene-7,19-dioic acid 7-methyl ester 19,2-lactone (**59**) [19 mg; contaminated with traces of (**28**)] (Found: M^+ , 388.1901. $C_{22}H_{28}O_6$ requires M , 388.1901); δ 1.07 (3 H, t, J 7.3 Hz, 2- CH_2 Me), 1.20 (3 H, s, 18- H_3), 2.57 (1 H, d, J 5.9 Hz, 6-H), 3.22 (1 H, dd, J 6.0 and 2.6 Hz, 5-H), 3.99 (1 H, br d, J 6 Hz, 3 α -H), 4.83 (br s, OH), 4.98 (1 H, br s, 17-H), 5.13 (1 H, br s, 17-H), and 5.57 (1 H, br s, 1-H) (on addition of D_2O the signal at 3.99 p.p.m. collapsed to a sharp singlet); m/z 388 (M^+ , 17%), 357 (14), 356 (33), 329 (13), 328 (17), 326 (39), 325 (100), 284 (19), 283 (32), 267 (50), 266 (44), 265 (46), 237 (18), and 71 (88); and a 1:1 mixture (39 mg) of 2-ethyl- GA_3 ethyl ester (**18**) and (**59**) (Found: M^+ , 388.1899. $C_{22}H_{28}O_6$ requires M , 388.1886); δ [spectrum of (**59**) subtracted] 1.04 (3 H, t, J 7.3 Hz, 2- CH_2 Me), 1.24 (3 H, s, 18- H_3), 2.77 (1 H, d, J 10.7 Hz, 6-H), 3.18 (1 H, d, J 10.7 Hz,

5-H), 3.74 (3 H, s, OMe), 3.99 (1 H, s, 3 α -H), 4.96 (1 H, br s, 17-H), 5.28 (1 H, br s, 17-H), and 5.98 (1 H, br s, 1-H).

(e) 2α -Ethyl- 1β -iodogibberellin A_1 methyl ester (**35**). The iodo lactone (**35**) (30 mg) and DBU (40 μ l) were refluxed in toluene (15 ml) for 1 h. Flash chromatography of the product, recovered as in (a), with ethyl acetate–light petroleum (1:1) yielded only starting material, identified by 1H n.m.r. and t.l.c.

Hydrolysis of 2-Methyl- GA_7 Methyl Ester (16) via the 3-Tetrahydropyranyl Ether.—2-Methyl- GA_7 methyl ester (**16**) (178 mg) was stirred in dry dichloromethane with 2,3-dihydropyran (150 μ l) and toluene- p -sulphonic acid (catalytic). After 1.5 h the mixture was evaporated under reduced pressure and the resultant gum purified by flash chromatography with ethyl acetate–light petroleum (1:3). The product in methanol (10 ml) was refluxed for 16 h with aqueous 2M sodium hydroxide (20 ml). Work-up gave a gum which was stirred in acetone (27 ml) and methanol (3 ml) containing toluene- p -sulphonic acid (catalytic) for 1.5 h. Evaporation under reduced pressure followed by flash chromatography with ethyl acetate–light petroleum–acetic acid (30:70:1) gave the 2β -methyl 19,2-lactone (**60**), identical by 1H n.m.r. and t.l.c. with that described earlier.

Reaction of 2-Methyl- GA_7 Methyl Ester (16) with Toluene- p -sulphonic Acid in Acetone–Methanol.—2-Methyl- GA_7 methyl ester (**16**) in acetone (13 ml) and methanol (2 ml) was stirred with toluene- p -sulphonic acid (catalytic). After 7 h, work-up gave ent- 2β - 3α -dihydroxy- 2α -methyl-20-norgibberella-1(10),16-diene-7,19-dioic acid 19,2-lactone 7-methyl ester (**61**) as a gum (Found: M^+ , 358.1812. $C_{21}H_{26}O_5$ requires M , 358.1780); δ 1.20 (3 H, s, 18- H_3), 1.58 (3 H, s, 2β -Me), 2.54 (1 H, d, J 6.2 Hz, 6-H), 3.20 (1 H, dd, J 6.2 and 2.5 Hz, 5-H), 3.73 (3 H, s, OMe), 3.89 (1 H, s, 3 α -H), 4.90 (2 H, br s, 17- H_2), and 5.51 (1 H, m, 1-H); m/z 358 (M^+ , 3%), 326 (15), 298 (15), 295 (100), 253 (29), 237 (50), 236 (31), 235 (41), 169 (19), 105 (16), 94 (46), and 91 (27).

Acknowledgements

We thank the S.E.R.C. for a research studentship (to A. M. F.), South Glamorgan Education Authority for a University grant (to D. N. M. J.) and the A.R.C. for a research grant. We also thank Dr. K. A. G. MacNeil for the low and high resolution probe mass spectra, and Messrs. P. Gaskin and M. J. Lee for the g.l.c.–mass spectra.

References

- 1 A. M. Fowles and J. MacMillan, *J. Chem. Soc., Perkin Trans. 1*, 1988, preceding paper.
- 2 J. MacMillan and D. A. Taylor, *J. Chem. Soc., Perkin Trans. 1*, 1985, 837.
- 3 M. H. Beale, *J. Chem. Soc., Perkin Trans. 1*, 1985, 1151.
- 4 M. H. Beale and J. MacMillan, *Phytochemistry*, 1981, **20**, 693.
- 5 J. MacMillan and C. L. Willis, *J. Chem. Soc., Perkin Trans. 1*, 1984, 351.
- 6 D. A. Taylor, Ph.D. Thesis, University of Bristol, 1983.
- 7 G. Costa, A. Carnus, L. Gatti, and N. Marsich, *J. Org. Chem.*, 1966, **5**, 568.
- 8 S. C. Dolan and J. MacMillan, *J. Chem. Soc., Chem. Commun.*, 1985, 1588.
- 9 P. S. Kirkwood, J. MacMillan, and M. L. Sinnott, *J. Chem. Soc., Perkin Trans. 1*, 1980, 2117.
- 10 W. G. Kofron and L. M. Baclawski, *J. Org. Chem.*, 1976, **41**, 1879.